

Table II
Chemical Resistance of Fortiflex HDPE

Legend: S = Satisfactory, O = Some Attack, U = Unsatisfactory

Reagent	21°C (70°F)	60°C (140°F)	Reagent	21°C (70°F)	60°F (140°F)
Copper Chloride Sat'd	S	S	Glycolic Acid 30%	S	S
Copper Cyanide Sat'd	S	S	Grape Juice	S	S
Copper Fluoride 2%	S	S	Grapefruit Juice	S	S
Copper Nitrate Sat'd	S	S	Heptane	O	U
Copper Sulfate Dilute	S	S	Hexachlorobenzene	S	S
Copper Sulfate Sat'd	S	S	Hexane	U	U
Corn Oil	S	S	Hydrobromic Acid 50%	S	S
Cottonseed Oil	S	S	Hydrocyanic Acid Sat'd	S	S
Cranberry Sauce	S	S	Hydrochloric Acid 10%	S	S
Cresols	S	O	Hydrochloric Acid 30%	S	S
Cuprous Chloride Sat'd	S	S	Hydrochloric Acid 35%	S	S
Cuprous Oxide	S	S	Hydrocyanic Acid	S	S
Cyclohexane	U	U	Hydrofluoric Acid 40%	S	S
Cyclonexanone	U	U	Hydrofluoric Acid 60%	S	S
Decalin	S	U	Hydrofluoric Acid 75%	S	S
Detergents Synthetic	S	S	Hydrogen 100%	S	S
Developers Photographic	S	S	Hydrogen Bromide 10%	S	S
Dextrin Sat'd	S	S	Hydrogen Chloride Gas Dry	S	S
Dextrose Sat'd	S	S	Hydrogen Peroxide 30%	S	S
Dibutyl Ether	O	U	Hydrogen Peroxide 90%	S	S
Dichlorobenzene (O&P)	U	U	Hydroquinone	S	S
Diethylene Glycol	S	S	Hydrogen Sulfide	S	S
Sodium Phosphate	S	S	Hypochlorous Acid Conc Irris	S	S
Dioxane	S	S	Iodine Crystals	O	O
Emulsions Photographic	S	S	Isobutyl Alcohol	S	S
Ether	O	O	Isopropyl Alcohol	S	S
Ethyl Acetate 100%	O	O	Isopropyl Ether	O	O
Ethyl Alcohol 100%	S	S	Kerosene	O	O
Ethyl Alcohol 35%	S	S	Lactic Acid 10%	S	S
Ethylbenzene	O	U	Lactic Acid 90%	S	S
Ethylene Glycol	S	S	Lanolin	S	S
Ferric Chloride Sat'd	S	S	Lard	S	S
Ferric Nitrate Sat'd	S	S	Lead Acetate Sat'd	S	S
Ferrous Ammonium Citrate	S	S	Lead Nitrate	S	S
Ferrous Chloride Sat'd	S	S	Lemon Juice	S	S
Ferrous Sulfate	S	S	Lemon Oil	O	U
Fluoboric Acid	S	S	Lime Juice	S	S
Fluorine	S	U	Linseed Oil	S	S
Fluosilicic Acid 32%	S	S	Magnesium Carbonate Sat'd	S	S
Fluosilicic Acid Conc.	S	S	Magnesium Chloride Sat'd	S	S
Formaldehyde			Magnesium Hydroxide Sat'd	S	S
10-30%	S	S	Magnesium Nitrate Sat'd	S	S
30-40%	S	O	Magnesium Sulfate Sat'd	S	S
Formic Acid 20%	S	S	Margarine	S	S
Formic Acid 50%	S	S	Mercuric Chloride	S	S
Formic Acid 100%	S	S	Mercuric Cyanide Sat'd	S	S
Fructose Sat'd	S	S	Mercurous Nitrate Sat'd	S	S
Fuel Oil	S	U	Mercury	S	S
Furfural 100%	O	U	Methyl Alcohol 100%	S	S
Furfuryl Alcohol	S	O	Methyl Ethyl Ketone 100%	U	U
Gallic Acid Sat'd	S	S	Methylsulfuric Acid	S	U
Gasoline	S	U	Methylene Chloride 100%	U	U
Glucose	S	S			
Glycerine	S	S			
Glycol	S	S			

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Reagent	21°C (70°F)	60°C (140°F)	Reagent	21°C (70°F)	60°F (140°F)
Milk	S	S	Tin	S	S
Mineral Oils	S	U	Zinc	S	S
Molasses	S	S	Potassium Bicarbonate Sat'd	S	S
Mustard (Prepared)	S	S	Potassium Borate 1%	S	S
Naphtha	O	U	Potassium Bromate 10%	S	S
Naphthalene	S	U	Potassium Bromide Sat'd	S	S
Natural Gas (Wet)	S	S	Potassium Carbonate	S	S
Nickel Chloride Sat'd	S	S	Potassium Chlorate Sat'd	S	S
Nickel Nitrate Conc.	S	S	Potassium Chloride Sat'd	S	S
Nickel Sulfate	S	S	Potassium Chromate 40%	S	S
Nicotinic Acid	S	S	Potassium Cyanide Sat'd	S	S
Nitric Acid 0-30%	S	S	Potassium Dichromate 40%	S	S
Nitric Acid 30-50%	S	O	Potassium Ferri/Ferro Cyanide	S	S
Nitric Acid 70%	S	O	Potassium Fluoride	S	S
Nitric Acid 95-98%	U	U	Potassium Hydroxide 20% Conc.	S	S
Nitrobenzene 100%	U	U	Potassium Nitrate Sat'd	S	S
Nitroglycerine	O	U	Potassium Perborate Sat'd	S	S
Octane	S	S	Potassium Perchlorate 10%	S	S
Oleum Conc.	U	U	Potassium Permanganate 20%	S	S
Olive Oil	S	S	Potassium Sulfate Conc.	S	S
Orange Juice	S	S	Potassium Sulfide Conc.	S	S
Oxalic Acid Dilute	S	S	Potassium Sulfite Conc.	S	S
Oxalic Acid Sat'd	S	S	Potassium Persulfate Sat'd	S	S
Ozone	O	O	Propane Gas	S	S
Palm Oil	S	S	Propargyl Alcohol	S	S
Paraffin Oil	S	O	Propyl Alcohol	S	S
Peanut Butter	S	S	Propylene Glycol	S	S
Perchloroethylene	U	U	Pyridine	S	O
Pepper (Fresh Ground)	U	S	Rayon Coagulating Bath	S	S
Peppermint Oil	O	U	Resorcinol	S	S
Perchloric Acid 50%	S	O	Salicylic Acid	S	S
Petroleum Ether	U	U	Sea Water	S	S
Petroleum Jelly	S	S	Shortening	S	S
Phenol	S	S	Silicic Acid	S	S
Phosphoric Acid 0-30%	S	S	Silver Nitrate Sol'n	S	S
Phosphoric Acid 30-90%	S	S	Soap Solution Conc.	S	S
Phosphoric Acid Over 90%	S	S	Sodium Acetate Sat'd	S	S
Photographic Solutions	S	S	Sodium Benzoate 35%	S	S
Phthalic Anhydride	S	S	Sodium Bicarbonate Sat'd	S	S
Pickling Baths	S	S	Sodium Bisulfate Sat'd	S	S
Sulfuric Acid	S	S	Sodium Bisulfite Sat'd	S	S
Hydrochloric Acid	S	S	Sodium Borate	S	S
Sulfuric-Nitric	S	U	Sodium Carbonate Conc.	S	S
Pine Oil	O	U	Sodium Chlorate Sat'd	S	S
Plating Solutions	S	S	Sodium Chloride Sat'd	S	S
Brass	S	S	Sodium Cyanide	S	S
Cadmium	S	S	Sodium Dichromate Sat'd	S	S
Chromium	S	S	Sodium Ferricyanide Sat'd	S	S
Copper	S	S	Sodium Ferricyanide	S	S
Gold	S	S	Sodium Fluoride Sat'd	S	S
Indium	S	S	Sodium Hydroxide Conc.	S	S
Lead	S	S			
Nickel	S	S			
Rhodium	S	S			
Silver	S	S			

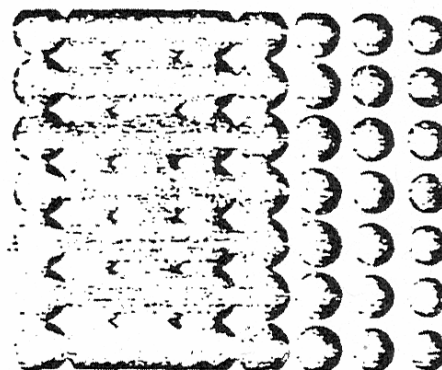
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Reagent	21°C (70°F)	60°C (140°F)	Reagent	21°C (70°F)	60°F (140°F)
Sodium Hypochlorite	S	S	Toluene	U	U
Sodium Nitrate	S	S	Tomato Juice	S	S
Sodium Nitrite	S	S	Transformer Oil	S	O
Sodium Perborate	S	S	Trisodium Phosphate Sat'd	S	S
Sodium Phosphate	S	S	Trichloroethylene	U	U
Sodium Sulfide 25% to Sat'd	S	S	Turpentine	O	U
Sodium Sulfite Sat'd	S	S	Urea	S	S
Sodium Thiosulphate	S	S	Urine	S	S
Soybean Oil	S	S	Vanilla Extract	S	S
Stannous Chloride Sat'd	S	S	Vaseline	S	S
Stannic Chloride Sat'd	S	S	Vinegar Comm.	S	S
Starch Solution Sat'd	S	S	Wetting Agents	S	S
Stearic Acid 100%	S	S	Whiskey	S	S
Sulfuric Acid 0-50%	S	S	Wines	S	S
Sulfuric Acid 70%	S	O	Xylene	U	U
Sulfuric Acid 80%	S	U	Yeast	S	S
Sulfuric Acid 96%	O	U	Zinc Chloride Sat'd	S	S
Sulfuric Acid 98% Conc	O	U	Zinc Oxide	S	S
Sulfuric Acid Fuming	U	U	Zinc Sulfate Sat'd	S	S
Sulfurous Acid	S	S			
Tannic Acid	S	S			
Tannic Acid 10%	S	S			
Tea	S	S			
Tetrahydrofuran	O	O			

Rigidex

high density
polyethylene



Technigram T800/1

Chemical resistance of Rigidex

Introduction

A guide to the effect that a variety of chemicals has on high density polyethylene (HDPE) is given in the following Table which is a compendium of BP Chemicals Limited information and other published literature sources.

Polyethylene materials as a class have a high resistance to chemical attack and are insoluble in all known solvents at room temperature. However, when considering the suitability of Rigidex HDPE for a particular application, certain other factors should be taken into account.

Permeation effects

HDPE has a closely packed molecular structure and as a result provides an effective barrier to many liquids, gases and vapours. Consequently, Rigidex grades are used extensively to package and transport a wide range of household and industrial products, foodstuffs, pharmaceuticals, agrochemicals and cosmetics. However, HDPE, in common with all thermoplastics is incapable of providing a complete barrier to all substances and loss by relatively low permeation rates can sometimes be accompanied by partial collapse of the container (panelling). In some instances, this can influence the stacking performance of the container in addition to detracting from the aesthetic appeal.

If panelling cannot be overcome by improving bottle rigidity, then it might sometimes be possible to camouflage the effect by using an oval cross section bottle or incorporating concave surfaces.

Photo-degradation effects

Panelling can also occur with products that are either photo-degradable or result in the absorption of oxygen from the ullage space. However, the use of opaque pigmentation and/or flushing the head space with nitrogen immediately prior to capping, will often eliminate panelling.

Absorption

This phenomenon is similar to permeation and in general the two effects tend to present problems with the same liquids. One notable exception is motor oil, which often exhibits quite high absorption levels yet negligible permeation.

When HDPE absorbs a liquid, there may be a change of dimensions and swelling. In severe cases, buckling of the container walls and panelling effects similar to those observed by permeation can occur. The swelling is usually accompanied by some loss of mechanical properties, notably tensile strength and softening of the polymer.

Environmental stress crack resistance

Products made from HDPE sometimes fail prematurely in service in the presence of certain surface active environments. Examples of these are detergents, chlorinated hydrocarbons, alcohols and some silicone products.

The brittle type of failure which occurs is always associated with a combination of the active environment and mechanical stresses in the moulded product. In any moulding there will always be regions of moulded-in stress in addition to stresses which result from externally applied loads. Also, some features of container design such as sharp corners or notches will produce areas of stress concentration which intensify the effects of applied loads. These stresses can result in brittle failure in the presence of active environments, even though the stresses involved would normally be within the mechanical limits of the polymer. In addition, these effects can be accelerated by elevated temperatures.

The activity of a liquid as a stress cracking environment and the resistance to stress cracking provided by the polymer may be assessed by a range of test methods. Generally, polyethylene materials of lower density and melt flow rate have better resistance to environmental stress cracking.

Packaging suitability

The use of small scale accelerated tests should be used only as an indication of the suitability of HDPE as a packaging material for a particular product. Information inferred from these tables for a single product may not, of course, be relevant when it is used as a component in a mixture of chemicals. Consequently, potential users of HDPE containers should always satisfy themselves that the containers are suitable for their particular product, taking into account the conditions of storage and distribution.

Guide to chemical resistance

Key

S. Satisfactory. This chemical is only absorbed to a low level and thus has little or no measurable effect on physical properties.

L. Limited resistance. A higher level of absorption occurs resulting in definite loss of physical properties. The question of the suitability of high density polyethylene would have to be considered with respect to the particular environment.

E. Environmental stress cracking may occur.

NS. Not satisfactory. Chemical attack or a high level of absorption occurs. In both cases the loss of physical properties is such that high density polyethylene is not suitable where prolonged contact is likely.

Substance	Conc. %	20°C	60°C
General chemical products			
Acetaldehyde	100	S	L
Acetic acid	10	S/E	S/E
Acetic acid	100	S/E	L/E
Acetic anhydride	100	S	L
Acetone	100	S	S
Acetophenetidin		S	S
Acetophenone	100	S	S
Acetylene	All	S	S
Acrylic emulsions	All	S/E	S/E
Acrylonitrile	100	S	S
Adipic acid		S	S
Aliphatic alcohols	100	S	S
Allyl chloride	100	S	S

Substance	Conc. %	20°C	60°C
Allyl chloride	100	L	L
Aluminium acetate	All	S	S
Aluminium carbonate	All	S	S
Aluminium chloride		S	S
Aluminium fluoride		S	S
Aluminium hydroxide		S	S
Aluminium nitrate		S	S
Aluminium oxalate		S	S
Aluminium oxychloride		S	S
Aluminium sulphate		S	S
Alums	All	S	S
Amino acetic acid	All	S	S
Ammonia (gaseous)	100	S	S
Ammonium acetate	All	S	S
Ammonium bicarbonate	All	S	S
Ammonium carbonate		S	S
Ammonium chloride		S	S
Ammonium fluoride		S	S
Ammonium hydroxide		S	S
Ammonium metaphosphate		S	S
Ammonium nitrate		S	S
Ammonium oxalate		S	S
Ammonium persulphate		S	S
Ammonium phosphate		S	S
Ammonium sulphate		S	S
Ammonium sulphide		S	S
Ammonium thiocyanate		S	S

Substance	Conc. %	20°C	60°C	Substance	Conc. %	20°C	60°C
Amyl alcohol	10	L	L	Chloroethanol	100	S	S
Amyl chloride	All	L	L	Chloroform	100	NS	NS
Aniline	100	L	L	Chloromethane	100	L	—
Antimony	100	S	S	Chlorosulphonic acid	100	NS	NS
Antimony trichloride	100	S	S	Chrome alum		S	S
Aqua regia	100	NS	NS	Chromic acid	Sat.	S/L	NS
Arsenic	100	S	S	Chromium salts	Sat.	S	S
Arsenic acid		S	S	Citric acid		S/E	S/E
Aryl sulphonic acid	100	S	S	Copper salts (aqueous)	Sat.	S	S
Benzaldehyde	100	S	L	Cresols	100	S	S
Benzaldehyde (AQ)	Sat.	S	—	Cresylic acid	100	L	—
Benzene	100	L	NS	Cresylic acid	50	S	S
Benzene sulphonic acid	100	S/E	S/E	Cyclohexane	100	L	L
Benzoic acid	Sat.	S	S	Cyclohexanol	100	S/E	S/E
Benzophenone	100	S	S	Cyclohexanone	100	S	L
Benzoyl chloride	100	L	L	Decahydronaphthalene	100	L	NS
Bismuth carbonate		S	S	Dibutyl ether	100	L	NS
Boric acid		S	S	Dibutyl phthalate	100	L	L
Boron trifluoride	100	S		Dibutyl sebacate	100	S	—
Bromic acid	10	S	S	Dichloroacetic acid	100	S	L
Bromine (liquid)	100	NS	NS	Dichloroacetic acid	50	S	S
Bromine (vapour)	High	NS	NS	Dichloroacetic acid/methyl ester		S	S
Bromine water	Sat.	NS	NS	o-dichlorobenzene		L	NS
Bromochloromethane	100	NS	NS	p-dichlorobenzene		L	NS
Butadiene	100	NS	NS	Dichloroethane	100	L	L
Butanediol	100	S/E		Dichloroethylene	100	NS	NS
Butane (gaseous)	100	S	S	Diethyl ether	100	L	L
Butane (liquid)	100	L	—	Diethylene glycol	100	S/E	S/E
Butanol	100	S	S	Diglycolic acid	100	S/E	S/E
Butoxyl (methoxy butyl acetate)	100	S	L	Dihexyl phthalate	100	S	S
Butyl acetate	100	L	NS	Diisobutyl ketone	100	S	NS
Butyl alcohol	100	S/E	S/E	Dimethyl carbinol	100	S	—
Butylene glycol	100	S	S	Dimethyl formamide	100	S	L
Butyric acid	100	S	L	Dimethyl sulphoxide	100	S	S
Carbon dioxide (dry)	100	S	S	Dinonyl adipate	100	S	—
Carbon dioxide (wet)	100	S	S	Dinonyl phthalate	100	S	L
Carbon disulphide	100	L	—	Diocyl adipate	100	S	—
Carbon monoxide	100	S	S	Diocyl phthalate	100	S	L
Carbon tetrachloride	100	NS	NS	1,4-Dioxane	100	S	S
Caustic potash (soln.)	50	S	S	Diphenyl ether	100	L	L
Caustic potash (soln.)	10	S	S	Disodium phosphate	100	S	S
Caustic soda	50	S	S	Epichlorhydrin	100	S	S
Caustic soda	10	S	S	Epoxy resins	100	S	S
Chloral hydrate	100	S	S	Ethanediol	100	S	S
Chlorine gas (moist)		L	NS	Ether	100	S/E	L
Chlorine liquid		NS	NS	Ethyl acetate	100	S	L
Chlorine water		L	NS	Ethyl acrylate	100	L	L
Chloroacetic acid	100	S	S	Ethyl alcohol	100	S/E	S/E
Chlorobenzene	100	L	NS	Ethylbenzene	100	L	NS

Substance	Conc. %	20°C	60°C	Substance	Conc. %	20°C	60°C
Ethyl butyrate	100	L	NS	Iron salts (aqueous)	Sat.	S	S
Ethyl chloride	100	L	NS	Isobutyl alcohol	100	S	—
Ethylene chloride	100	L	NS	Isooctane	100	S	L
Ethylenediaminetetraacetic acid	100	S	S	Isopropanol	100	S	S
Ethylene dichloride	100	L	NS	Isopropyl ether	100	L	NS
Ethylene glycol	100	S/E	S/E	Lactic acid	100	S/E	S/E
2-Ethylhexanol	100	S	—	Lead acetate	Sat.	S	S
Fatty acids (> 6)		S/E	S/E	Lead tetraethyl		S	
Ferric chloride		S	S	Magnesium carbonate		S	S
Ferric nitrate		S	S	Magnesium chloride		S	S
Ferric sulphate		S	S	Magnesium hydroxide		S	S
Ferrous ammonium citrate chloride	Sat.	S	S	Magnesium nitrate		S	S
Ferrous sulphate		S	S	Magnesium sulphate		S	S
Fluoboric acid	100	S	S	Maleic acid	50	S	S
Fluoric acid	40	S	S	Manganese sulphate		S	S
Fluorine	100	NS	NS	Mercuric chloride		S	S
Fluorosilicic acid		S	S	Mercuric cyanide		S	S
Formaldehyde (aqueous)	40	S/E	S/E	Mercurous nitrate		S	S
Formalin		S	S	Mercury	100	S	S
Formic acid		S/E	S/E	Methoxybutanol	100	S	L
Furfural	100	L	NS	Methyl acrylate		NS	NS
Furfuryl alcohol	100	S	L	Methyl alcohol		S/E	S/E
Gallic acid		S/E	S/E	Methyl bromide		L	NS
Glycolic acid	30	S/E	S/E	Methyl cyclohexane		L	NS
Glycolic acid butyl ester	100	S	S	Methylene chloride		NS	NS
Heptane	100	S	NS	Methyl ethyl ketone	100	S	L
Hexachlorobenzene	100	S	S	Methyl glycol		S	S
Hexamine	100	S/E	S/E	4-Methyl-2-pentanol		S	L
Hexane	100	NS	NS	Methyl sulphuric acid		S/E	S/E
Hydrazine	100	S/E	S	Monochloroacetic acid		S	S
Hydrazine hydrate	100	S/E	S	Monochlorobenzene		S	S
Hydrobromic acid	50	S	S	Naphthalene	100	S	
Hydrochloric acid	Conc.	S	S	Natural gas		S	S
Hydrocyanic acid	100	S	S	Nickel salts (aqueous)	Sat.	S	S
Hydrocyanic acid	Sat.	S	S	Nicotine	Dil.	S/E	S/E
Hydrocyanic acid	10	S	S	Nicotinic acid		S/E	S/E
Hydrofluoric acid	40	S	S	Nitric acid	50	L	NS
Hydrofluoric acid	70	S	S	Nitric acid	98	NS	NS
Hydrogen	100	S	S	Nitrobenzene	100	NS	NS
Hydrogen bromide	10	S	S	Nitroglycerine	100	NS	NS
Hydrogen peroxide	30	S	S	Nitropropane		L	L
Hydrogen peroxide	90	S	NS	o-nitrotoluene	100	S	L
Hydrogen phosphite	100	S	S	Nitrous gases	100	S	S
Hydrogen sulphide	Low	S	S	Nonyl alcohol	100	S	—
Hydroquinone		S	S	Oleic acid	100	S/E	L/E
Hypochlorites	100	S	S	Oleum		NS	NS
Hypochlorous acid	All	S	S	Oxalic acid		S/E	S/E

Substance	Conc. %	20°C	60°C	Substance	Conc. %	20°C	60°C
Oxygen	100	S	S	Potassium perborate		S	S
Ozone		NS	NS	Potassium permanganate		S	S
Palmitic acid		S/E	S/E	Potassium persulphate		S	S
Pentane		NS	—	Potassium sulphate		S	S
Perchloric acid	20	S	S	Potassium sulphide		S	S
Perchloric acid	50	S	S	Potassium sulphite		S	S
Perchloric acid	70	S	NS	Propane (gaseous)	100	S	S
Petroleum ether	100	S	L	Propane (liquid)	100	S	—
Phenol (aqueous phase)		S	S	Propargyl alcohol		S/E	S/E
Phenol (solid phase)	100	S	S	Propionic acid	50	S	S
Phenyl sulphonate		S	S	Propionic acid	100	S	L
Phosgene		S	—	Propyl alcohol	100	S/E	S/E
Phosphoric acid	25	S	S	Propylene dichloride	100	NS	NS
Phosphoric acid	50	S	S	Propylene glycol		S/E	S/E
Phosphoric acid	Conc.	S/L	NS	Pseudocumene		L	L
Phosphoric anhydride		S	S	Pyridine	100	S	NS
Phosphorous		S	S	Quinine		S	
Phosphorous oxychloride		S	L	Resorcinol	100	S	S
Phosphorous pentoxide		S	S	Salicylic acid		S	S
Phosphorous trichloride		S	L	Selenic acid	100	S	S
Phthalic acid	50	S	S	Silicic acid		S	S
Phthalic anhydride		S	S	Silicone fluids		S/E	S/E
Picric acid		S	S	Silver nitrate		S	S
Polyglycol ethers		S	S	Silver salts (aqueous)	Sat.	S	S
Potassium antimonate		S	S	Sodium acetate		S	S
Potassium bicarbonate		S	S	Sodium benzoate		S	S
Potassium bichromate		S	S	Sodium bicarbonate		S	S
Potassium bisulphate		S	S	Sodium bisulphate		S	S
Potassium borate		S	S	Sodium borate		S	S
Potassium bromate		S	S	Sodium bromide		S	S
Potassium bromide		S	S	Sodium carbonate		S	S
Potassium carbonate		S	S	Sodium chlorate		S	S
Potassium chlorate		S	S	Sodium chloride		S	S
Potassium chloride		S	S	Sodium chlorite		S	S
Potassium chromate		S	S	Sodium cyanide		S	S
Potassium cuprocyanide		S	S	Sodium dichromate		S	S
Potassium cyanide		S	S	Sodium dodecylbenzenesulphonate		S	S
Potassium dichromate		S	S	Sodium ferricyanide		S	S
Potassium ferri/ferrocyanide		S	S	Sodium ferrocyanide		S	S
Potassium fluoride		S	S	Sodium fluoride		S	S
Potassium hydroxide		S	S	Sodium hydroxide		S	S
Potassium hypochlorite		S	S	Sodium hypochlorite		S	S
Potassium iodide	Sat.	S	S	Sodium hyposulphate		S	—
Potassium nitrate		S	S	Sodium nitrate		S	S
Potassium orthophosphate		S	S	Sodium nitrite		S	S

Substance	Conc. %	20°C	60°C
Sodium orthophosphate		S	S
Sodium perborate		S	S
Sodium peroxide	10	S	—
Sodium peroxide	Sat.	L	—
Sodium phosphate		S	S
Sodium silicate		S	S
Sodium sulphate		S	S
Sodium sulphide		S	S
Sodium sulphite		S	S
Sodium tetraborate		S	S
Sodium thiosulphate		S	S
Stannic chloride		S	S
Stannous chloride		S	S
Stearic acid	100	S/E	S/E
Succinic acid	Sat.	S	S
Sulphur		S	S
Sulphur dioxide (dry)		S	S
Sulphur dioxide (wet)		S	S
Sulphur trioxide		NS	NS
Sulphuric acid	10	S	S
Sulphuric acid	50	S	S
Sulphuric acid	95	S	L
Sulphurous acid	30	S	S
Sulphuryl chloride		NS	
Tannic acid		S/E	S/E
Tartaric acid	100	S	S
Tartaric acid (aqueous)	Sat.	S	S
Tetrabromomethane	100	NS	NS
Tetrachloroethane	100	NS	NS
Tetrahydrofuran	100	L	NS
Tetralin	100	S	L
Thioglycolic acid	100	S	S
Thionyl chloride		NS	
Thiophene	100	L	L
Toluene	100	L	NS
Trichloroacetic acid	50	S	S
Trichloroacetic acid	100	L	NS
Trichloroethylene	100	L	NS
Tri & chloroethylene phosphate	100	S	S
Tricresyl phosphate	100	L	L
Triethanolamine		L	L
Urea		S/E	S/E

Substance	Conc. %	20°C	60°C
Xylene	100	L	NS
Zinc ammonium carbonate		S	S
Zinc carbonate		S	S
Zinc chloride		S	S
Zinc oxide		S	S
Zinc sulphate		S	S

Oils and Waxes

Aniseed oil	100	L	NS
Beeswax	100	S	S
Carnauba wax		S	S
Castor oil	100	S/E	S/E
Cedar leaf oil		NS	NS
Cedar wood oil		NS	NS
Cinnamon oil		NS	NS
Citronella oil		NS	NS
Clove oil		S	S
Coconut oil		S	NS
Coconut oil alcohols		S/E	S/E
Cod liver oil	100	S	—
Corn seed oil		S	L
Cotton seed oil		S/E	S/E
Diesel oil		S	S
Fuel oil	100	S	L
Kerosene		L	L
Lanolin	100	S	—
Lemon peel oil		S	L
Linseed oil	100	S	S
Menthol	100	S	S
Mineral oil		L	L
Motor oil	100	S	L
Olive oil	100	S	S
Orange peel oil	100	S	—
Palm oil	100	S	—
Paraffin oil	100	S	L
Paraffin wax	100	S	S
Peanut oil	100	S	S
Peppermint oil		L	NS
Petrol		S/L	S/L
Pine needle oil	100	S	S
Pine oil	100	L	NS
Salad oils	100	S	L
Silicone oil	100	S/E	L/E

Substance	Conc.%	20°C	60°C	Substance	Conc.%	20°C	60°C
Soya bean oil	100	S	S	Pectin	Sat	S	S
Tallow	100	S	—	Pepper		S	S
Transformer oil	100	S	L	Pineapple juice		S	S
Turpentine oil	100	L	NS	Rum		S	S
White spirit		S/L	L	Saccharose	100	S	S
Foods etc.				Soda water		S	—
Beer		S	S	Starch		S	S
Brandy		S	—	Sugar beet syrup	100	S	S
Butter		S	—	Tea (leaves)		S	S
Buttermilk		S	—	Tomato juice	100	S	S
Cider		S/E	S/E	Tomato ketchup	100	S	S
Cinnamon		S	—	Vanilla	100	S	S
Cloves		S	—	Vinegar	All	S	S
Coca Cola		S	S	Whisky	40	S/E	
Cocoa		S	—	Wine (mulled claret)	100	S	S
Coffee		S	—	Wine spirit	100	S	S
Cream		S	—	Yeast	100	S	S
Curds		S	—				
Dextrose		S	S	Car, garden, household products			
Fructose		S	S	Antifreeze		S	S
Fruit juices		S	S	Bleach		L	L
Gelatine		S	S	Borax		S	S
Gin	40	S	L	Brake fluid	100	S/E	S/E
Glucose		S	S	Brine		S	S
Glycerine	100	S/E	S/E	Creosote	100	S	S
Glycerine (aqueous)	High	S	S/E	Cresol (aqueous)		S	S
Glycerine (aqueous)	Low	S	S	Detergents		S/E	S/E
Grapefruit juice		S	S	Dextron		S	S
Honey		S	S	Emulsions (photographic)	100	S/E	S/E
Horse radish		S	—	Floor wax	100	S	L
Jam	100	S	S	Furniture polish	100	S/E	L
Jelly	100	S	S	Hair shampoo		S	S
Lard		S		Hydraulic fluid	100	S	L
Lemon juice		S	S	Ink	100	S	S
Lime juice		S	S	Methylated spirit		S/E	S/E
Margarine	100	S	S	Nail polish and remover	100	S	L
Mayonnaise	100	S	—	Petroleum jelly	100	S	S
Milk		S	S	Shoe polish	100	S	L
Molasses		S	S	Talc		S	S
Mustard		S	S	Tar	100	S	L
Orange juice		S	S				
Paprika	100	S	S				